S.2 CHEMISTRY END OF TEM ONE 2023 CHEMISTRY ASSESSMENT EXAM

Name:	•••••
School:	•••••

TIME: 1Hour: 30 minutes

Uganda Certificate of Education

Chemistry Paper 1

INSTRUCTIONS:

- This paper consists of **four** questions.
- Answer all questions in the spaces provided.

Illustrations in form of drawings should be made where necessary, with a sharp pencil

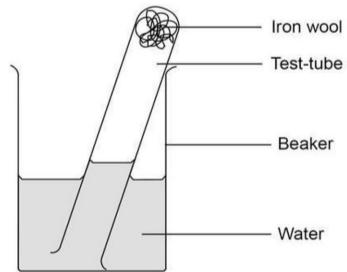
Table for Examiners use only

	Marks	Comment.
Number		
1		
2		
3		
4		

1. Objects we use in everyday life are made from materials which	ch are either
plastics, ceramics; metals; wood, metals or glass.	
(a)State two useful properties of plastics.	(02 marks)
(b)Plastic waste management is a challenge in our communities disposal of plastics as shown in the picture below is a big threat around the globe.	
(a)What threats do plastics pose to the environment and man.	(05 marks)

	•••••
	•••••
(b)Suggest two possible solutions to the threats plastics pose to the	
environment you have written in (a). (02 mar)	ks)
2. Below is a flow used summarized by one group of S.2 students about n terms of elements; compounds and mixtures.	natter in
terms of elements, compounds and mixtures.	
No Does it have constant Yes	
properties and composition? Mixture Pure substance	
No Is it uniform Yes No Can it be simplified Yes	_
throughout? chemically?	
Heterogeneous Element Compou	and
(a)Classify the following substances as element; compound or mixture.	
(i) Sugar. (01 r	nark)
(i) Charcoal. (01 r	mark)

(ii)	Sugar solution.	(01 mark)
•••••		
(b)State t	two differences in prop	perties between sugar and sugar solution.
		(02 marks)
•••••		
(c)Briefly	y describe how pure w	rater can be obtained from sugar solution.
		(03 marks)
(d)Name	the substance that can	be used to test for the pure water obtained in
		served when the substance named is used to
, ,	ne pure water obtained	
3. Below	is a set up apparatus t	hat was used to investigate the most active
		answer the questions that follow



(a)(i)Identify the component of air being investigated?	(01 mark)
(ii)State the approximate percentage of the component of the air being investigated in the experiment above.	g (01 mark)
	1 mark)
(iv)State the type of change that took place leading to the process nan (iii) above and give one reason for the type of change that took place.(02 mrks)
(b)State why there was rise of some water inside the test tube? (04	-marks)